



(12) **EUROPEAN PATENT APPLICATION**

(43) Date of publication:
25.06.2003 Bulletin 2003/26

(51) Int Cl.⁷: **B41M 5/00, G03C 1/76**

(21) Application number: **02080095.9**

(22) Date of filing: **09.12.2002**

(84) Designated Contracting States:
AT BE BG CH CY CZ DE DK EE ES FI FR GB GR
IE IT LI LU MC NL PT SE SI SK TR
Designated Extension States:
AL LT LV MK RO

(30) Priority: **20.12.2001 US 28130**
15.10.2002 US 271057

(71) Applicant: **EASTMAN KODAK COMPANY**
Rochester, New York 14650 (US)

(72) Inventors:
• **Landry-Coltrain, Christine J.T.**
Rochester, New York 14650-2201 (US)

• **Leon, Jeffrey W.**
Rochester, New York 14650-2201 (US)
• **Franklin, Linda M.**
Rochester, New York 14650-2201 (US)
• **Yacobucci, Paul Daniel**
Rochester, New York 14650-2201 (US)

(74) Representative: **Honoré, Anne-Claire et al**
Kodak Industrie
Département Brevets
C.R.T. 60/2
Zone Industrielle
71102 Chalon-sur-Saône Cedex (FR)

(54) **Porous organic particles for ink recording element use**

(57) The present invention comprises porous polyester particles comprising porous polyester particles having a mean diameter of less than 0.5 micrometers. The present invention further comprises an ink recording element comprising a support having thereon at least one ink receiving layer capable of accepting an ink image, said layer(s) comprising porous polyester particles comprising porous polyester particles having a

mean diameter of less than 0.5 micrometers. The present invention also includes a method of forming an ink print comprising providing an ink recording element comprising porous polyester particles comprising porous polyester particles having a mean diameter of less than 0.5 micrometers and printing on said ink recording element utilizing an ink printer.

Description

[0001] This invention relates to porous organic particles. The invention further relates to an ink recording element, more particularly to an inkjet recording element containing porous organic particles.

[0002] In a typical inkjet recording or printing system, ink droplets are ejected from a nozzle at high speed towards a recording element or medium to produce an image on the medium. The ink droplets, or recording liquid, generally comprise a recording agent, such as a dye or pigment, and a large amount of solvent. The solvent, or carrier liquid, typically is made up of water, an organic material such as a monohydric alcohol, a polyhydric alcohol or mixtures thereof.

[0003] An inkjet recording element typically comprises a support having on at least one surface thereof an ink-receiving or image-forming layer and includes those intended for reflection viewing, which have an opaque support, and those intended for viewing by transmitted light, which have a transparent support.

[0004] An inkjet recording element that simultaneously provides an almost instantaneous ink dry time and good image quality is desirable. However, given the wide range of ink compositions and ink volumes that a recording element needs to accommodate, these requirements of inkjet recording media are difficult to achieve simultaneously.

[0005] Inkjet recording elements are known that employ porous or non-porous single layer or multilayer coatings that act as suitable ink receiving layers on one or both sides of a porous or non-porous support. Recording elements that use non-porous coatings typically have good image quality and stability but exhibit poor ink dry time. Recording elements that use porous coatings typically contain colloidal particulates and have poorer image stability but exhibit superior dry times.

[0006] While a wide variety of different types of porous image recording elements for use with inkjet printing are known, there are many unsolved problems in the art and many deficiencies in the known products which have severely limited their commercial usefulness. A major challenge in the design of a porous image-recording layer is to be able to obtain good quality, crack-free coatings. Inkjet prints, prepared by printing onto inkjet recording elements, are subject to environmental degradation. They are especially vulnerable to light fade and fade resulting from gaseous impurities in the air, such as ozone and nitrous oxide. Highly swellable hydrophilic layers can take an undesirably long time to dry, slowing printing speed. Porous layers speed the absorption of the ink vehicle, but often suffer from insufficient gloss and severe dye fade. Porous layers are also difficult to coat without cracking.

[0007] Japanese Kokai 07-137432 describes an inkjet paper having an ink-absorbing layer containing polyester resin particles with internal pores. However, there is a problem with this element in that the average particle size of the polyester resin is greater than 0.5 microns, and the element will have low surface gloss.

[0008] It is an object of this invention to provide an ink recording element which will provide improved ink uptake speed. Another objective of the invention is to provide an ink recording element having high surface gloss. Another objective of the invention is to provide an ink recording element having a receiving layer that when printed upon has an excellent image quality and stability.

[0009] The present invention comprises an imaging element comprising at least one layer comprising porous organic particles, wherein said porous organic particles comprise an unsaturated condensation polymer reacted with a vinyl monomer. The present invention further comprises porous polyester particles comprising porous polyester particles having a mean diameter of less than 0.5 micrometers. The present invention further comprises an ink recording element comprising a support having thereon at least one ink receiving layer capable of accepting an ink image, said layer(s) comprising porous polyester particles comprising porous polyester particles having a mean diameter of less than 0.5 micrometers. The present invention also includes a method of forming an ink print comprising providing an ink recording element comprising porous polyester particles comprising porous polyester particles having a mean diameter of less than 0.5 micrometers and printing on said ink recording element utilizing an ink printer.

[0010] Using the invention, a recording element is obtained which will provide improved ink uptake speed, high surface gloss and, when printed upon, has an excellent image quality.

[0011] The present invention details the use of porous organic particles in an ink recording element, wherein the porous organic particles comprise an unsaturated condensation polymer reacted with a vinyl monomer. Porous, condensation polymer particles may be prepared by crosslinking an unsaturated precursor condensation polyester within an oil-in-water emulsion in the presence of a water-immiscible organic solvent. The crosslinked, porous condensation polymer particles may be prepared via methods which are analogous to those described below for porous polyester particles with the main difference being that an unsaturated precursor condensation polymer is used in lieu of an unsaturated precursor polyester. The precursor condensation polymer is a polymer containing a backbone consisting of repetitive organic diradicals linked together by one or more of the following bond types: amide, carbonate, urethane, ester, or urea bonds. Preferably, the precursor condensation polymer will contain ester bonds and one or more of non-ester bond types. For example, the unsaturated condensation polymer may comprise at least one of ester-co-urethane, ester-co-urea, ester-co-amide, or ester-co-carbonate, most preferably ester-co-urethane or ester-co-carbonate. In one embodiment, the porous organic particles comprise an unsaturated condensation polymer reacted with a vinyl monomer such as styrene, divinylbenzene, divinyl adipate, and cyclohexanedimethanol divinyl ether. The polymer may be linear

or branched.

[0012] The precursor condensation polymer may also contain chemical unsaturation through which it can be crosslinked within an oil-in-water emulsion in the presence of a water-immiscible organic liquid to afford porous particles. The chemical unsaturation may be present within the precursor polyester along the backbone, as functionalized end groups or as pendant groups. An example of the first case is a polyester-urethane of which one of the repetitive ester units is a maleate or fumarate moiety. An example of the second case is an alcohol-terminated polyurethane which has been reacted with methacryloyl chloride to afford methacrylate ester end groups. Preferably the chemical unsaturation will be present as backbone unsaturation.

[0013] The precursor condensation polymers may be synthesized using any of the techniques commonly known to those skilled in the art of polymer synthesis for preparing condensation polymers. In general, the methods involve the reaction of lewis acidic and lewis basic monomers, each with a functionalization number of two or more under solution or melt conditions. Specific reagent combinations are shown in Table 7. It should be noted that multifunctional reagents with functionality numbers other than 2 (for example, trifunctional, tetrafunctional) may also be used. Conditions may be chosen in which the different types of reagents will react in a single reactor. Alternately, a multiple stage approach may be used in which a prepolymer, macromonomer, or oligomer with appropriate terminating groups is reacted with one or more additional polyfunctional reagents in a subsequent step. For example, a polyester-carbonate may be prepared either by reacting a diacid chloride, a bischloroformate, and a diol in the same pot or by preparing a low molecular weight alcohol-terminated prepolymer, which is subsequently reacted with a bischloroformate.

Table 7.

Reagent combinations (difunctional cases) required for forming condensation polymers.		
<u>Condensation polymer bond</u>	<u>Lewis acid</u>	<u>Lewis base</u>
Ester	Cyclic anhydride, diacid chloride, diester, diacid	Diol, diphenol
Urethane	Diisocyanate	Diol, diphenol
Urea	Diisocyanate, phosgene or derivative thereof	Diamine
Carbonate	Bischloroformate, phosgene or reactive derivative thereof	Diol, diphenol
Amide	Cyclic anhydride, diacid chloride, diester, diacid	Diamine

[0014] The crosslinking reaction is a radical-initiated polymerization of an ethylenically unsaturated monomer which readily copolymerizes with the unsaturated units in the precursor condensation polymer. The precursor condensation polymer can be organic-soluble, in which case an added emulsifying agent may be necessary. In another embodiment of this method, the precursor condensation polymer can be water-soluble, water-dispersible, or amphiphilic in character, in which case the precursor condensation polymer acts as the emulsifying species and an added emulsifying agent is merely optional. The methods by which the porous condensation polymer particles may be prepared as well as the other reagents used in the preparation (i.e. emulsifiers, initiators, ethylenically unsaturated monomers, water-immiscible organic) are the same as those described below for the porous polyester particles. Preferably, the porous, condensation polymer particles will contain ionic groups, as described below for porous polyester particles. Preferably, these ionic groups will be quaternary ammonium moieties.

[0015] The porous, condensation polymer beads may have a mean diameter of 0.1 - 10 micron. Preferably, they will have a mean diameter of 0.1-0.5 micron.

[0016] The most preferred porous organic particles useful for this invention are described in Serial Number 10/027,701 by Leon et al., (Docket 82842) entitled "Method of Preparation of Porous Polyester Particles". The crosslinked, porous polyester particles may be prepared by crosslinking an unsaturated precursor polyester within an oil-in-water emulsion in the presence of a water-immiscible organic solvent. A precursor polyester is a polyester containing unsaturated groups which is used in turn to make porous polyester particles. The crosslinking reaction is a radical-initiated polymerization of an ethylenically unsaturated monomer which readily copolymerizes with the unsaturated units in the precursor polyester. The precursor polyester can be organic-soluble, in which case an added emulsifying agent may be necessary. In another embodiment of this method, the precursor polyester can be water-soluble, water-dispersible, or amphiphilic in character, in which case the precursor polyester acts as the emulsifying species and an added emulsifying agent is merely optional. The water-immiscible organic solvent is removed to yield a dispersion of porous, crosslinked, polyester-containing particles.

[0017] The precursor polyesters which may be used to form the porous polyester particles useful for this invention may be branched or unbranched, contain chemical unsaturation, and may be soluble either in water-immiscible organic solvents or in water. Optionally, the precursor polyester may be self-emulsifying in water or amphiphilic or surfactant-

like in character. The precursor polyesters may have any glass transition temperature, provided it fulfills the solubility requirements. Preferably, the number average molecular weight (M_n) is from 1,000 to 30,000 gm/mole.

[0018] As is well known in the art, polyesters are condensation products of polybasic acids or of corresponding acid equivalent derivatives such as esters, anhydrides or acid chlorides and polyhydric alcohols. It will be known that whenever "diacids" or "polyacids" are referred to in this document, that corresponding acid equivalent derivatives such as esters, anhydrides or acid chlorides are also included by reference. Polymerizable unsaturation may be introduced into the molecule by the selection of a polybasic acid or polyhydric alcohol, which contains α,β -ethylenic unsaturation. In most cases, the unsaturation will be contained within the polybasic acid unit. Optionally, one or more additional polyacids common in the art of polycondensation may be used in addition to the unsaturated polyacid. These ethylenically unsaturated polyacids include, but are not necessarily limited to maleic, fumaric, itaconic, phenylenediacrylic, citraconic and mesaconic acid. Other, additional polyacids which do not contain chemical unsaturation and can be used in polyesters are described in WO 01/00703. These diacids can include, but are not necessarily limited to malonic, succinic, glutaric, adipic, pimelic, azelaic, and sebacic acids, phthalic, isophthalic, terephthalic, tetrachlorophthalic, tetrahydrophthalic, trimellitic, trimesic, isomers of naphthalenedicarboxylic acid, chlorendic acid, trimellitic acid, trimesic acid, and pyromellitic acid.

[0019] Ethylenically unsaturated groups can also be introduced into the precursor polyester by synthetic modification. For example, a polyester with a high alcohol number can be reacted with an anhydride or acid chloride of acrylic acid or methacrylic acid in order to introduce ethylenically unsaturated units.

[0020] Precursor polyesters that may be suitable for this invention can furthermore be comprised of any of a wide variety of polyhydric alcohols which are well known in the art of polycondensation and may be aliphatic, alicyclic, or aralkyl. A description of suitable polyhydric alcohols is given in WO 01/00703. These alcohols can include, but are not necessarily limited to ethylene glycol, 1,3-propylene glycol, 1,6-hexanediol, 1,10-decanediol, 1,4-cyclohexanedimethanol, 1,4-cyclohexanediol, hydroquinone bis (hydroxyethyl) ether, diethylene glycol, neopentyl glycol, bisphenols such as bisphenol A, ethylene oxide and propylene oxide adducts of bisphenol A, pentaerythritol, trimethylolpropane, and polyester polyols, such as that obtained by the ring-opening polymerization of ϵ -caprolactone. Additionally, A-B type polycondensation monomers which contain both hydroxyl and acid derivative functions can be used as well as monoacids and monoalcohols.

[0021] In one embodiment of this invention, precursor polyesters which are water-soluble, surfactant-like, or self-emulsifying and additionally contain chemical unsaturation may be utilized. Water-soluble, surfactant-like, and self-emulsifying polyesters are well known in the art and may contain one or more type of hydrophilic chemical group, functionality, or monomer, such as carboxylate, quaternary ammonium, sulfonate, sulfate, sulfonium, phosphonium, iminosulfonyl, or polymeric or oligomeric oxyethylene segments. Precursor polyesters used to form the porous polyester particles useful in this invention may additionally contain one or more polyacid or polyol monomers which contain ethylenic unsaturation as detailed above. The water-soluble, surfactant-like, and self-emulsifying precursor polyesters used to form the porous polyester particles useful in this invention may contain one or more diacid or diol components which can induce hydrophilic character or water-solubility. The most common hydrophilic diol used for this purpose is polyethylene glycol. Additionally, tertiary amine units substituted with two or three hydroxyalkyl groups can be incorporated within a precursor polyester and rendered ionic either by quaternization with an alkylating agent or by neutralization with an acid. A commonly used class of diacid components used to impart hydrophilicity to polyesters includes compounds containing sulfonate or sulfonimide salts. Some suitable sulfonated diacids are described in U.S. patents 4,973,656 and 5,218,042. Examples of such diacids are 5-sodiosulfoisophthalic acid, 2-sodiosulfobutanoic acid, and di-Me sodioiminobis(sulfonyl-*m*-benzoate). Another common strategy for the hydrophilization of polyesters involves the neutralization of the acid end groups of polyester with a relatively high acid number. Preferably, the acid number is at least 10. Most preferably the acid number is greater than 25. The neutralization agent is usually an alkali metal hydroxide or an amine. Polyesters containing ethylenic unsaturation and neutralized acid end groups can also be used in this invention. In the preferred case, the unsaturated precursor polyester will contain an ionic group equivalent weight of from 400 to 2000 grams of polymer per mole of ionic unit.

[0022] The ethylenically unsaturated monomers useful for crosslinking the precursor polyesters in this invention may be monomers commonly used in the art of addition polymerization. These include, but are not necessarily limited to methacrylic acid esters, such as methyl methacrylate, ethyl methacrylate, isobutyl methacrylate, 2-ethylhexyl methacrylate, benzyl methacrylate, phenoxyethyl methacrylate, cyclohexyl methacrylate and glycidyl methacrylate, acrylate esters such as methyl acrylate, ethyl acrylate, isobutyl acrylate, 2-ethylhexyl acrylate, benzyl methacrylate, phenoxyethyl acrylate, cyclohexyl acrylate, and glycidyl acrylate, styrenics such as styrene, α -methylstyrene, 3- and 4-chloromethylstyrene, halogen-substituted styrenes, and alkyl-substituted styrenes, vinyl halides and vinylidene halides, N-alkylated acrylamides and methacrylamides, vinyl esters such as vinyl acetate and vinyl benzoate, vinyl ethers, such as butyl vinyl ether and cyclohexanedimethanol divinyl ether, allyl alcohol and its ethers and esters, and unsaturated ketones and aldehydes such as acrolein and methyl vinyl ketone and acrylonitrile.

[0023] In addition, small amounts (typically less than 10% of the total weight of the polymerizable solids) of one or

more water-soluble ethylenically unsaturated monomer can be used. Such monomers include but are not necessarily limited to styrenics, acrylates, and methacrylates substituted with highly polar groups, unsaturated carbon and heteroatom acids such as acrylic acid, methacrylic acid, fumaric acid, maleic acid, itaconic acid, vinylsulfonic acid, vinylphosphonic acid, and their salts, vinylcarbazole, vinylimidazole, vinylpyrrolidone, and vinylpyridines.

[0024] Especially useful in this invention are monomers containing more than one ethylenically unsaturated unit, such as trimethylolpropane triacrylate, ethylene glycol dimethacrylate, isomers of divinylbenzene, divinyl adipate, cyclohexanedimethanol divinyl ether and ethylene glycol divinyl ether.

[0025] Ethylenically unsaturated monomers which are preferred for this invention are styrenics, vinyl ethers, and methacrylates. Divinylbenzene (m, and p isomers), styrene, divinyl adipate, and ethylene glycol dimethacrylate are especially preferred.

[0026] Any of the common water-soluble or organic-soluble free radical polymerization initiators known in the art of addition polymerization can be used for this invention. These include, but are not restricted to azo compounds, such as 2,2'-azobis(4-methoxy-2,4-dimethylvaleronitrile), (1-phenylethyl)azodiphenylmethane, 2,2'-azoisobutyronitrile (AIBN), 1,1'-azobis(1-cyclohexanedicarbonitrile), 4,4'-azobis(4-cyanopentanoic acid), and 2,2'-azobis(2-amidinopropane) dihydrochloride, organic peroxides, organic hydroperoxides, peresters, and peracids such as benzoyl peroxide, lauryl peroxide, capryl peroxide, acetyl peroxide, t-butyl hydroperoxide, t-butyl perbenzoate, cumyl hydroperoxide, peracetic acid, 2,5-dimethyl-2,5-di(peroxybenzoate), and p-chlorobenzoyl peroxide, persulfate salts such as potassium, sodium and ammonium persulfate, disulfides, tetrazenes, and redox initiator systems such as $\text{H}_2\text{O}_2/\text{Fe}^{2+}$, persulfate/bisulfite, oxalic acid/ Mn^{3+} , thiourea/ Fe^{3+} , and benzoyl peroxide/dimethylaniline.

[0027] Optionally, a small amount of a cosurfactant stabilizer, typically comprising 1-10% by weight of the organic phase, may be added to the organic phase of this invention. These hydrophobic compounds are known to prevent Ostwald ripening in certain types of emulsion and suspension polymerization. Excellent discussions of cosurfactants are given in *Emulsion Polymerization and Emulsion Polymers* by Peter A Lovell and Mohammed S. El-Aaser, (John Wiley and Sons: Chichester, 1997; pp. 700-721) and US Patent 5,858,634. The most common cosurfactants are hexadecane and hexadecanol. Other useful cosurfactants may also serve other roles, such as acting as monomers or initiators. An example of the former is lauryl methacrylate. An example of the latter is lauroyl peroxide.

[0028] If a precursor polyester is used in this invention which is not soluble or dispersible in water, then an emulsifier may additionally be used, although an emulsifier can be used in tandem with a water-soluble or water-dispersible precursor polyester. It may be preferable that the emulsifier be present in the aqueous phase. Though a very large variety of emulsifiers are known in the art, most of these fit into the three basic categories of surfactants, colloidal inorganics, and protective colloids. There exist a tremendous number of known surfactants. Good reference sources for surfactants are the *Surfactant Handbook* (GPO: Washington, D. C., 1971) and *McCutcheon's Emulsifiers and Detergents* (Manufacturing Confectioner Publishing Company: Glen Rock, 1992). There are no general restrictions for the surfactants which may be useful in this invention. Useful surfactants can be anionic, cationic, zwitterionic, neutral, low molecular weight, macromolecular, synthetic, or extracted or derived from natural sources. Some examples include, but are not necessarily limited to: sodium dodecylsulfate, sodium dodecylbenzenesulfonate, sulfosuccinate esters, such as those sold under the AEROSOL® trade name, fluorosurfactants, such as those sold under the ZONYL® and FLUORAD® trade names, ethoxylated alkylphenols, such as TRITON® X-100 and TRITON® X-705, ethoxylated alkylphenol sulfates, such as RHODAPEX® CO-436, phosphate ester surfactants such as GAFAC® RE-90, hexadecyltrimethylammonium bromide, polyoxyethylenated long-chain amines and their quaternized derivatives, ethoxylated silicones, alkanolamine condensates, polyethylene oxide-co-polypropylene oxide block copolymers, such as those sold under the PLURONIC® and TECTRONIC® trade names, N-alkylbetaines, N-alkyl amine oxides, and fluorocarbon-poly(ethylene oxide) block surfactants, such as FLUORAD® FC-430.

[0029] Protective colloids useful in this invention include, but are not necessarily limited to: poly (ethylene oxide), hydroxyethyl cellulose, poly (vinyl alcohol), poly (vinyl pyrrolidone), polyacrylamides, polymethacrylamides, sulfonated polystyrenes, alginates, carboxy methyl cellulose, polymers and copolymers of dimethylaminoethylmethacrylate, water soluble complex resinous amine condensation products of ethylene oxide, urea and formaldehyde, polyethyleneimine, casein, gelatin, albumin, gluten and xanthan gum. Protective colloids are a class of emulsifiers which may be used in lieu of or in addition to a surfactant. They may be typically dissolved or dispersed in the aqueous phase prior to the emulsification step.

[0030] Similarly, colloidal inorganic particles can be employed as emulsifiers as part of a limited coalescence process. Colloidal inorganic particles can be employed in lieu of or in addition to any other type of emulsifier listed, such as a surfactant or protective colloid. They may be also added to the aqueous phase. Limited coalescence techniques have been describe in numerous patents such as U.S. Patents 4,833,060 and 4,965,131. A colloidal inorganic which is particularly useful in this invention is LUDOX® TM sold by Du Pont.

[0031] Additional additives which can be incorporated into the porous organic particles useful in this invention include pigments, dyes, biocides, fungicides, electrolytes, buffers, UV-absorbers, antioxidants and chain transfer agents.

[0032] The porous polyester particles useful for this invention comprise porous polyester particles having a mean

diameter of less than 0.5 micrometers. For optimal ink absorption properties and coating quality, it may be preferable that the porous polyester particles have a mean diameter range from 0.1 to less than 0.5 micrometers, and more preferably, that the porous polyester particles have a mean diameter range from 0.2 to 0.3 micrometers. The diameter of the particles can be measured by any method known in the art. One such method may be laser light scattering of dilute dispersions of the particles, using a commercially available instrument such as the Horiba LA-920, manufactured by Horiba LTD. Typically, a sample of porous polyester particles will contain a population of particles having a distribution of sizes. This is the particle size distribution, and is characterized by a mean diameter, a standard deviation, and a coefficient of variation. The mathematical equations defining these terms can be found in any basic text on statistical analysis, such as "Principles of Instrumental Analysis, 4th Edition", by D. A. Skoog and J. J. Leary, Harcourt Brace College Publishers, Orlando, FL, 1971 (Appendix A-6). The mean diameter is the arithmetic mean of the particle size distribution. The coefficient of variation (CV) of a distribution is the ratio of the standard deviation of the distribution to the mean diameter, given as a percent. The porous polyester particles useful for this invention can have a relatively large distribution of particle sizes within one mode, and the standard deviation in the mean diameter can be from 0.3 times the mean particle diameter to 3 times the mean particle diameter. In a system of particles, there can be a single mode or peak to this distribution of sizes, or there can be several modes, each mode being characterized by a mean diameter, a standard deviation, and a coefficient of variation. For example, the porous polyester particles can be a system composed of particles having a mode with a mean diameter of less than 0.5 micrometers and particles having a mode with mean diameter greater than 0.5 micrometers, preferably having a mean diameter from 1 to 10 micrometers, and most preferably having a mean diameter from 1 to 3 micrometers. The relative proportions of these two modes are calculated from the relative areas under the curves representing the modes, and should add up to 100 %.

[0033] As discussed above, the porous condensation polymer particles preferably contain ionic groups. In the most preferred case, the porous polyester particles will contain ionic groups. Preferably the particles will have an ionic group equivalent weight of from 40 to 2000 grams per mole of ionic unit. These ionic groups may be ammonium (primary, secondary, tertiary, or quaternary), pyridinium, imidazolium, alkylsulfonates, alkylthiosulfate, carboxylate, phosphonium or sulfonium. Copolymerizable, α , (β -ethylenically unsaturated monomers containing a preformed ionic functionality can be used in any of the polymerization processes described herein. Suitable monomers which can be used include, for example, the following monomers and their mixtures: cationic ethylenically unsaturated monomers, for example, vinylbenzyltrimethylammonium chloride, vinylbenzyl-dimethyl-dodecylammonium chloride, other vinylbenzylammonium salts in which the three other ligands on the nitrogen can be any alkyl or carbocyclic group including cyclic amines such as piperidine, the counter ions of which can be halides, sulfonates, phosphates, sulfates, [2-(methacryloyloxy)ethyl]trimethyl-ammonium chloride, [2-(acryloyloxy)ethyl]-trimethylammonium p-toluene-sulfonate, and other acrylate and methacrylate ammonium salts in which the alkyl group connecting the acrylic function to the nitrogen can be ≥ 2 carbon atoms long and the other three nitrogen ligands can be any alkyl or carbocyclic group including cyclic amines such as piperidine, and benzyl, 4-vinyl-1-methylpyridinium methyl sulfate, 3-methyl-1-vinylimidazolium methosulfate, and other vinylpyridinium and vinylimidazolium salts in which the other nitrogen ligand may be any alkyl or cycloalkyl group, vinyltriphenylphosphonium bromide, vinylbenzyltriphenylphosphonium tosylate, and other phosphonium salts in which the other three phosphorous ligands may be any aromatic or alkyl group. In a preferred embodiment, the cationic functionality may be vinylbenzyltrimethylammonium chloride, vinylbenzyl-N-butylimidazolium chloride, vinylbenzyl-dimethyldodecylammonium chloride or vinylbenzyl-dimethyloctadecylammonium chloride.

[0034] Other suitable copolymerizable, α , β -ethylenically unsaturated monomers containing a preformed ionic functionality which can be used include, for example, the following monomers and their mixtures: anionic ethylenically unsaturated monomers such as 2-phosphatoethyl acrylate potassium salt, 3-phosphatopropyl methacrylate ammonium salt, and other acrylic and methacrylic esters of alkylphosphonates in which the alkyl group connecting the acrylic function to the phosphate function can be ≥ 2 carbon atoms long, the counter ions of which can be alkali metal cations, quaternary ammonium cations, phosphonium cations, or the like, sodium methacrylate, potassium acrylate, and other salts of carboxylic acids, styrenesulfonic acid ammonium salt, methyltriphenylphosphonium styrenesulfonate, and other styrene sulfonic acid salts, 2-sulfoethyl methacrylate pyridinium salt, 3-sulfopropyl acrylate lithium salt, and other acrylic and methacrylic esters of alkylsulfonates, and other sulfonates such as ethylene sulfonic acid sodium salt. In a preferred embodiment, the anionic functionality may be trimethylamine hydrochloride salt of methacrylic acid, dimethylbenzylamine hydrochloride salt of methacrylic acid, dimethyldodecyl-amine hydrochloride salt of methacrylic acid or methyltriocetylammmonium salt of styrenesulfonic acid.

[0035] The ionic group can also be formed after the polymer particle is prepared by modifying non-ionic monomers to make them (or part of them) ionic. All of the cationic and anionic functionalities mentioned above can be incorporated by modifying a non-ionic polymer particle.

[0036] The product particles, having excellent colloidal stability, can be stored as an aqueous dispersion or freeze dried to yield a solid powder comprising dry particles which will easily redisperse in water.

[0037] The ink receiving layer of the ink element may be formed by coating a mixture comprised of these porous organic particles and a binder in an amount insufficient to alter the porosity of the porous receiving layer onto a support,

and then drying to remove approximately all of the volatile components. In a preferred embodiment, the polymeric binder is a hydrophilic polymer such as polyvinylpyrrolidone and vinylpyrrolidone-containing copolymers, polyethyloxazoline and oxazoline-containing copolymers, imidazole-containing polymers, polyacrylamides and acrylamide-containing copolymers, poly(vinyl alcohol) and vinyl-alcohol-containing copolymers, poly(vinyl methyl ether), poly(vinyl ethyl ether), poly(alkylene oxides), gelatin, cellulose ethers, poly(vinylacetamides), partially hydrolyzed poly(vinyl acetate/vinyl alcohol), poly(acrylic acid), sulfonated or phosphated polyesters and polystyrenes, casein, albumin, chitin, chitosan, dextran, pectin, collagen derivatives, collodian, agar-agar, arrowroot, guar, carrageenan, tragacanth, xanthan, rhamosan. In another preferred embodiment of the invention, the hydrophilic polymer is hydroxyethyl cellulose, hydroxypropyl cellulose, hydroxypropyl methyl cellulose, methyl cellulose, or a poly(alkylene oxide). In still another preferred embodiment, the polymeric binder may be a latex such as poly(styrene-co-butadiene), polyurethane, polyester, poly(acrylate), poly(methacrylate), a copolymer of n-butylacrylate and ethylacrylate, and a copolymer of vinylacetate and n-butylacrylate. In still another preferred embodiment, the polymeric binder may be a water dispersible condensation polymer such as a polyurethane. In still another preferred embodiment, the binder may be a condensate of alkoxysilanes or other metal sols such as alumina sol, titania sol, or zirconia sol. Mixtures of the above listed hydrophilic polymers can be used. The binder should be chosen so that it is compatible with the aforementioned particles.

[0038] The amount of polymer binder used should be sufficient to impart cohesive strength to the ink recording element, but should also be minimized so that the interconnected pore structure formed by the aggregates is not filled in by the binder. In a preferred embodiment of the invention, the porous organic particles are present in an amount of from 50 to 95 % by weight, and most preferably, in an amount from 75 to 90 % by weight of the layer.

[0039] Since the image recording element may come in contact with other image recording articles or the drive or transport mechanisms of image recording devices, additives such as filler particles, surfactants, lubricants, crosslinking agents, matte particles may be added to the element to the extent that they do not degrade the properties of interest.

[0040] Filler particles may be used in the ink receiving layer such as silicon oxide, fumed silica, silicon oxide dispersions such as those available from Nissan Chemical Industries and DuPont Corp., aluminum oxide, fumed alumina, calcium carbonate, barium sulfate, barium sulfate mixtures with zinc sulfide, inorganic powders such as γ -aluminum oxide, chromium oxide, iron oxide, tin oxide, doped tin oxide, alumino-silicate, titanium dioxide, silicon carbide, titanium carbide, and diamond in fine powder, as described in U.S. Patent 5,432,050.

[0041] A dispersing agent, or wetting agent can be present to facilitate the dispersion of the filler particles. This helps to minimize the agglomeration of the particles. Useful dispersing agents include, but are not limited to, fatty acid amines and commercially available wetting agents such as Solsperse® sold by Zeneca, Inc. (ICI). Preferred filler particles may be silicon oxide, aluminum oxide, calcium carbonate, and barium sulfate. Preferably, these filler particles have a median diameter less than 1.0 μm . The filler particles can be present in the amount from 0 to 80 percent of the total solids in the dried ink receiving layer, most preferably in the amount from 0 to 40 percent.

[0042] In order to obtain adequate coatability, rheology modifiers known to those familiar with such art such as thickening agents or polymers may be used. These include associative thickeners such as hydrophobically modified hydroxyethylcellulose, hydrophobically modified alkali-soluble or alkali swellable emulsions, and hydrophobically modified ethylene oxide urethane block copolymers such as those supplied by Rohm & Haas under the trade name of Acusol® and Dow Chemical under the trade name of Polyphobe®, and non-associative thickeners such as hydroxyethyl cellulose, hydroxypropyl methyl cellulose, sodium carboxy methyl cellulose, xanthan gum, guar gum, and carrageenan.

[0043] The ink element may include lubricating agents. Lubricants and waxes useful either in the ink receiving layer or on the side of the element that is opposite the ink receiving layer include, but are not limited to, polyethylenes, silicone waxes, natural waxes such as carnauba, polytetrafluoroethylene, fluorinated ethylene propylene, silicone oils such as polydimethylsiloxane, fluorinated silicones, functionalized silicones, stearates, polyvinylstearate, fatty acid salts, and perfluoroethers. Aqueous or non-aqueous dispersions of submicron size wax particles such as those offered commercially as dispersions of polyolefins, polypropylene, polyethylene, high density polyethylene, microcrystalline wax, paraffin, natural waxes such as carnauba wax, and synthetic waxes from such companies as, but not limited to, Chemical Corporation of America (Chemcor), Inc., Michelman Inc., Shamrock Technologies Inc., and Daniel Products Company, are useful.

[0044] In order to obtain adequate coatability, additives known to those familiar with such art such as surfactants, defoamers, alcohol may be used. Coating aids and surfactants include, but are not limited to, nonionic fluorinated alkyl esters such as FC-430®, FC-431®, FC-10®, FC-171® sold by Minnesota Mining and Manufacturing Co., Zonyl® fluorochemicals such as Zonyl-FSN®, Zonyl-FTS®, Zonyl-TBS®, Zonyl-BA® sold by DuPont Corp., other fluorinated polymer or copolymers such as Modiper F600® sold by NOF Corporation, polysiloxanes such as Dow Corning DC 1248®, DC200®, DC510®, DC 190® and BYK 320®, BYK 322®, sold by BYK Chemie and SF 1079®, SF1023®, SF 1054®, and SF 1080® sold by General Electric, and the Silwet® polymers sold by Union Carbide, polyoxyethylene-lauryl ether surfactants, sorbitan laurate, palmitate and stearates such as Span® surfactants sold by Aldrich, poly(oxyethylene-co-oxypropylene) surfactants such as the Pluronic® family sold by BASF, and other polyoxyethylene-containing surfactants such as the Triton X® family sold by Union Carbide, ionic surfactants, such as the Alkanol® series sold by

DuPont Corp., and the Dowfax® family sold by Dow Chemical. Specific examples are described in MCCUTCHEON's Volume 1: Emulsifiers and Detergents, 1995, North American Edition.

[0045] The ink receiving layer may include crosslinking agents. Any crosslinking agent may be used provided its reactive functionalities have the appropriate reactivity with specific chemical units in the binder. Some common crosslinkers which can crosslink binders rich in lewis basic functionalities include, but are not necessarily limited to: carbodiimides, polyvalent metal cations, organic isocyanates such as tetramethylene diisocyanate, hexamethylene diisocyanate, diisocyanato dimethylcyclohexane, dicyclohexylmethane diisocyanate, isophorone diisocyanate, dimethylbenzene diisocyanate, methylcyclohexylene diisocyanate, lysine diisocyanate, tolylene diisocyanate, diphenylmethane diisocyanate, aziridines such as taught in U. S. Patent 4,225,665, ethyleneimines such as Xama-7® sold by EIT Industries, blocked isocyanates such as CA BI-12 sold by Cytec Industries, melamines such as methoxymethylmelamine as taught in U. S. Patent 5,198,499, alkoxysilane coupling agents including those with epoxy, amine, hydroxyl, isocyanate, or vinyl functionality, Cymel® crosslinking agents such as Cymel 300®, Cymel 303®, Cymel 1170®, Cymel 1171® sold by Cytec Industries, and bis-epoxides such as the Epon® family sold by Shell. Other crosslinking agents include compounds such as aryloylureas, aldehydes, dialdehydes and blocked dialdehydes, chlorotriazines, carbamoyl pyridiniums, pyridinium ethers, formamidinium ethers, vinyl sulfones, boric acid, dihydroxydioxane, and polyfunctional aziridines such as CX-100 (manufactured by Zeneca Resins). Such crosslinking agents can be low molecular weight compounds or polymers, as discussed in U. S. Patent 4,161,407 and references cited.

[0046] To improve colorant fade, UV absorbers, radical quenchers or antioxidants may also be added to the ink-receiving layer as is well known in the art. Examples include polyalkylenepolyamine-dicyanodiamide based polycondensation products, water soluble reducing agents, such as sulfites, nitrites, phosphates, thiosulfates, ascorbic acid or salts thereof, hydroxylamine derivatives, and glucose, sulfur-containing compounds, such as thiocyanates, thiourea, 2-mercaptobenzimidazole, 2-mercaptobenzthiazole, 2-mercaptobenzoxazole, 5-mercapto-1-methyl-tetrazole, 2,5-dimercapto-1,3,4-triazole, 2,4,6-trimercaptocyanuric acid, thiosalicylic acid, thiouracil, 1,2-bis(2-hydroxyethylthio) ethane, or hydrophobic antioxidant emulsified dispersions, such as hindered phenol based antioxidants, piperidine based antioxidants or hindered amines. UV absorbers include those described in Japanese Patent Publication Open to Public Inspection Nos. 57-74193, 57-87988, and 2-261476, antifading agents include those described in Japanese Patent Publication Open to Public Inspection Nos. 57-74192, 57-87989, 60-72785, 61-146591, 1-95091, and 3-13376.

[0047] The ink receiving layer may include pH modifiers, adhesion promoters, rheology modifiers, latexes, biocides, dyes, optical brighteners, whitening agents, described in Japanese Patent Publication Open to Public Inspection Nos. 59-42993, 59-52689, 62-280069, 61-242871, and 4-219266, and antistatic agents.

[0048] The ink receiving layer of the invention can contain one or more mordanting species or polymers. The mordant polymer can be a soluble polymer, a charged molecule, or a crosslinked dispersed microparticle. The mordant can be non-ionic, cationic or anionic. Examples of a mordant are polymers or copolymers containing a quaternized nitrogen moiety, such as, for example, poly(styrene-co-1-vinylimidazole-co-1-vinyl-3-benzylimidazolium chloride), poly(styrene-co-1-vinylimidazole-co-1-vinyl-3-hydroxyethyl-imidazolium chloride), poly(styrene-co-1-vinylimidazole-co-1-vinyl-3-benzylimidazolium chloride-co-1-vinyl-3-hydroxyethylimidazolium chloride), poly(vinylbenzyltrimethylammonium chloride-co-divinylbenzene), poly(ethyl acrylate-co-1-vinylimidazole-co-1-vinyl-3-benzylimidazolium chloride), or poly(styrene-co-4-vinylpyridine-co-4-hydroxyethyl-1-vinylpyridinium chloride). In a preferred embodiment of the invention, the quaternary nitrogen moiety incorporated in the polymer may be a salt of trimethylvinylbenzylammonium, benzyldimethylvinylbenzylammonium, dimethyloctadecylvinylbenzylammonium, glycidyltrimethylammonium, 1-vinyl-3-benzylimidazolium, 1-vinyl-3-hydroxyethylimidazolium or 4-hydroxyethyl-1-vinylpyridinium. Preferred counter ions which can be used include chlorides or other counter ions as disclosed in U.S. Patents 5,223,338, 5,354,813, and 5,403,955. Other mordants suitable for the invention may be cationic modified products of polymers such as poly(vinyl alcohol), gelatin, chitosan, polyvinylamine, polyethylene-imine, polydimethyldiallyl ammonium chloride, polyalkylene-polyamine dicyandiamide ammonium condensate, polyvinylpyridinium halide, polymers of (meth)acryloyl oxyalkyl quaternary ammonium salt, polymers of (meth)acrylamide alkyl quaternary ammonium salt, ω -chloro-poly(oxyethylene-polymethylene quaternary ammonium alkylate), methyl glycol chitosan, poly(vinylpyridine), propylene oxide based triamines of the Jeffamine T series, made by Texaco, Inc., quaternary acrylic copolymer latexes, phosphonium compounds, sulfonimides, sulfonated polymers and dispersed particles, and alumina hydrate. Other mordants suitable for the invention may be polymers, copolymers, or latexes containing carboxylic acid, sulfonic acid, sulfonamide, sulfonimide, or phosphonic acid, such as carboxylated and sulfonated acrylates or methacrylates, carboxylated styrene butadienes, sulfonated nylons, polyesters and polyurethanes, and their salts. In a preferred embodiment of this invention, the mordanting unit may be chemically incorporated within the chemical structure of the polyester bead. For example, a sulfonated monomer within the porous polyester structure may serve as a mordant for cationic dye species. The ink receiving element may contain multiple individual ink receiving layers. Each being comprised of a different composition, combinations of porous organic particles with differing mean diameters, and layer thickness. For these multilayer structures, the terms as used herein, "top", "upper", and "above" mean the layer that is farther from the support in relation to the relative positioning with respect to the other layers. The terms "bottom" "lower" and "below" mean the layer that

is closer to the support in relation to the relative positioning with respect to the other layers. In one embodiment, the ink recording element has a layer structure wherein at least one layer comprising porous polyester particles having a mean diameter of greater than 0.5 micrometers is located below a layer comprising porous organic particles having a mean diameter of less than 0.5 micrometers. In another embodiment, the ink recording element has a structure with
 5 at least one layer of porous polyester particles, having a mean diameter of greater than 0.5 micrometers in combination with porous polyester particles having a mean diameter of less than 0.5 micrometers, located below a layer comprising porous organic particles having a mean diameter of less than 0.5 micrometers. In another embodiment, the ink recording element has a structure wherein at least one layer comprising porous polyester particles having a mean diameter of greater than 0.5 micrometers is located below a layer comprising porous organic particles, having a mean diameter of
 10 less than 0.5 micrometers in combination with porous polyester particles having a mean diameter of greater than 0.5 micrometers. In yet another embodiment, the ink recording element has a structure wherein at least one layer comprising porous polyester particles, having a mean diameter of greater than 0.5 micrometers in combination with porous polyester particles having a mean diameter of less than 0.5 micrometers, is located below a layer comprising porous organic particles, having a mean diameter of less than 0.5 micrometers in combination with porous polyester particles
 15 having a mean diameter of greater than 0.5 micrometers.

[0049] The total thickness of the ink receiving layer(s) may range from 5 to 100 μm , preferably from 10 to 50 μm . The coating thickness required is determined through the need for the coating to act as a sump for absorption of ink solvent and the need to hold the ink near the coating surface. In a multilayer structure, the ink receiving layer containing porous organic particles having a mean diameter of less than 0.5 micrometers has a preferred thickness of from 1 to
 20 20 micrometers. The ink recording element containing porous organic, especially polyester, particles having a mean diameter of greater than 0.5 micrometers has a preferred thickness of from 5 to 50 micrometers. The ink receiving layers have a preferred combined thickness of from 6 to 65 micrometers. In one embodiment of the invention, the ink recording element demonstrates improved ozone resistance when the element further comprises absorbed copper phthalocyanine dye, thereby having a dye density loss in ozone of less than 2 % per day per ppm ozone. When, the
 25 ink recording element is prepared as single layer, gloss improvements become most easily measurable and the element preferably demonstrates a surface gloss of greater than or equal to 20 at 60 degrees.

[0050] In addition to the ink receiving layer, the recording element may also contain a base layer, next to the support, the function of which is to absorb the solvent from the ink. Materials useful for this layer include inorganic particles and polymeric binder, or highly swellable polymers such as gelatin.

[0051] The support for the ink recording element used in the invention can be any of those usually used for inkjet receivers. The support can be either transparent or opaque. Opaque supports include plain paper, coated paper, res-
 30 incoated paper such as polyolefin-coated paper, synthetic paper, photographic paper support, melt-extrusion-coated paper, and polyolefin-laminated paper, such as biaxially oriented support laminates. Biaxially oriented support laminates are described in U.S. Patents 5,853,965, 5,866,282, 5,874,205, 5,888,643, 5,888,681, 5,888,683, and 5,888,714. These biaxially oriented supports include a paper base and a biaxially oriented polyolefin sheet, typically polypropylene,
 35 laminated to one or both sides of the paper base. The support can also consist of microporous materials such as polyethylene polymer-containing material sold by PPG Industries, Inc., Pittsburgh, Pennsylvania under the trade name of Teslin®, Tyvek® synthetic paper (DuPont Corp.), impregnated paper such as Duraform®, and OPPalyte® films (Mobil Chemical Co.) and other composite films listed in U.S. Patent 5,244,861. Transparent supports include glass,
 40 cellulose derivatives, such as a cellulose ester, cellulose triacetate, cellulose diacetate, cellulose acetate propionate, cellulose acetate butyrate, polyesters, such as poly(ethylene terephthalate), poly(ethylene naphthalate), poly-1,4-cyclohexanedimethylene terephthalate, poly(butylene terephthalate), and copolymers thereof, polyimides, polyamides, polycarbonates, polystyrene, polyolefins, such as polyethylene or polypropylene, polysulfones, polyacrylates, polyether imides, and mixtures thereof. The papers listed above include a broad range of papers, from high end papers, such
 45 as photographic paper to low end papers, such as newsprint. In a preferred embodiment, Ektacolor paper made by Eastman Kodak Co. is employed. The term as used herein, "transparent" means the ability to pass radiation without significant deviation or absorption.

[0052] The support used in the invention may have a thickness of from 50 to 500 μm , preferably from 75 to 300 μm . Antioxidants, brightening agents, antistatic agents, plasticizers and other known additives may be incorporated into
 50 the support, if desired.

[0053] In order to improve the adhesion of the ink receiving layer to the support, an under-coating or subbing layer may be applied to the surface of the support. This layer may be an adhesive layer such as, for example, halogenated phenols, partially hydrolyzed vinyl chloride-co-vinyl acetate polymer, vinylidene chloride-methyl acrylate-itaconic acid
 55 terpolymer, a vinylidene chlorideacrylonitrile-acrylic acid terpolymer, or a glycidyl (meth)acrylate polymer or copolymer. Other chemical adhesives, such as polymers, copolymers, reactive polymers or copolymers, that exhibit good bonding between the ink receiving layer and the support can be used. The polymeric binder in the subbing layer employed in the invention is preferably a water soluble or water dispersible polymer such as poly(vinyl alcohol), poly(vinyl pyrrolidone), gelatin, a cellulose ether, a poly(oxazoline), a poly(vinylacetamide), partially hydrolyzed poly(vinyl acetate/vinyl

alcohol), poly(acrylic acid), poly(acrylamide), poly(alkylene oxide), a sulfonated or phosphated polyester or polystyrene, casein, zein, albumin, chitin, chitosan, dextran, pectin, a collagen derivative, collodian, agar-agar, arrowroot, guar, carrageenan, tragacanth, xanthan, rhamsan, a latex such as poly(styrene-co-butadiene), a polyurethane latex, a polyester latex, or a poly(acrylate), poly(methacrylate), poly(acrylamide) or copolymers thereof.

5 [0054] In a preferred embodiment, the subbing layer polymeric binder is a sulfonated polyester dispersion, such as AQ29® (Eastman Chemical Co.), gelatin, a polyurethane or poly(vinyl pyrrolidone). The polymeric binder for the subbing layer is preferably used in an amount of from 1 to 50 g/m², preferably from 1 to 20 g/m².

10 [0055] A borate or borate derivative employed in the subbing layer of the ink jet recording element of the invention may be, for example, borax, sodium tetraborate, boric acid, phenyl boronic acid, or butyl boronic acid. As noted above, the borate or borate derivative may be used in an amount of from 3 to 50 g/m², preferably from 3 to 10 g/m². It is believed that upon coating, the borate or borate derivative in the subbing layer diffuses into the image-receiving layer to cross-link the cross-linkable binder in the image-receiving layer.

15 [0056] Other methods to improve the adhesion of the layer to the support include surface treatment of the support by corona-discharge, plasma-treatment in a variety of atmospheres, UV treatment, which may be performed prior to applying the layer to the support.

20 [0057] The recording element of the invention can contain one or more conducting layers such as an antistatic layer to prevent undesirable static discharges during manufacture and printing of the image. This may be added to either side of the element. Antistatic layers conventionally used for color films have been found to be satisfactory, such as those in U.S. Patent 5,147,768. Preferred antistatic agents include metal oxides, e.g., tin oxide, antimony doped tin oxide and vanadium pentoxide. These antistatic agents may be preferably dispersed in a film-forming binder.

25 [0058] The layers described above may be coated by conventional coating means onto a support material commonly used in this art. Coating methods may include, but are not limited to, wound wire rod coating, knife coating, slot coating, slide hopper coating, gravure coating, spin coating, dip coating, skim-pan-air-knife coating, multilayer slide bead, doctor blade coating, gravure coating, reverse-roll coating, curtain coating, multilayer curtain coating. Some of these methods allow for simultaneous coatings of more than one layer, which may be preferred from a manufacturing economic perspective if more than one layer or type of layer needs to be applied. Known coating and drying methods are described in further detail in Research Disclosure no. 308119, published Dec. 1989, pages 1007 to 1008. Slide coating may be preferred, in which several layers may be simultaneously applied. The support may be stationary, or may be moving so that the coated layer may be immediately drawn into drying chambers. After coating, the layers may be generally dried by simple evaporation, which may be accelerated by known techniques such as convection heating.

30 [0059] The coating composition may be applied to one or both substrate surfaces through conventional pre-metered or post-metered coating methods listed above. The choice of coating process would be determined from the economics of the operation and in turn, would determine the formulation specifications such as coating solids, coating viscosity, and coating speed. After coating, the ink recording element may be subject to calendaring or supercalendering to enhance surface smoothness. In a preferred embodiment of the invention, the ink recording element is subject to hot soft-nip calendaring at a temperature of 65°C and a pressure of 14000 kg/m at a speed of from 0.15 m/s to 0.3 m/s.

35 [0060] Inks used to image the recording elements of the present invention are well known in the art. The ink compositions used in inkjet printing typically may be liquid compositions comprising a solvent or carrier liquid, dyes or pigments, humectants, organic solvents, detergents, thickeners, preservatives. The solvent or carrier liquid can be solely water or can be water mixed with other water-miscible solvents such as polyhydric alcohols. Inks in which organic materials such as polyhydric alcohols may be the predominant carrier or solvent liquid may also be used. Particularly useful may be mixed solvents of water and polyhydric alcohols. The dyes used in such compositions may be typically water-soluble direct or acid type dyes. Such liquid compositions have been described extensively in the prior art including, for example, U.S. Patents 4,381,946, 4,239,543 and 4,781,758.

40 [0061] Used herein, the phrase "ink recording element", which may also be referred to as an "imaging element" comprises an imaging support as described above along with an image receiving or recording layer as applicable to multiple techniques governing the transfer of an image onto the imaging element. Such techniques include thermal dye transfer with thermosensitive imaging materials, electrophotographic printing, or inkjet printing, as well as a support for photographic silver halide images. As used herein, the phrase "photographic element" is a material that utilizes photosensitive silver halide in the formation of images. The stabilized particle of the present invention may be used in a single technique or may be used in a hybrid system combining one or more technique. An example of a hybrid system might be an inkjet printing application on a photographic element.

45 [0062] The thermal ink or dye image-receiving or recording layer of the receiving or recording elements of the invention may comprise, for example, a polycarbonate, a polyurethane, a polyester, polyvinyl chloride, poly(styrene-co-acrylonitrile), poly(caprolactone), or mixtures thereof. The ink or dye image-receiving or recording layer may be present in any amount that is effective for the intended purpose. An overcoat layer may be further coated over the ink or dye-receiving or recording layer, such as described in U.S. Patent No. 4,775,657 of Harrison et al.

50 [0063] Ink or dye-donor elements that may be used with the ink or dye-receiving or recording element of the invention

conventionally comprise a support having thereon an ink or dye containing layer. Any ink or dye can be used in the ink or dye-donor employed in the invention, provided it is transferable to the ink or dye-receiving or recording layer by the action of heat. Ink or dye donors applicable for use in the present invention are described, e.g., in U.S. Patent Nos. 4,916,112, 4,927,803, and 5,023,228. As noted above, ink or dye-donor elements may be used to form an ink or dye transfer image. Such a process comprises image-wise-heating an ink or dye-donor element and transferring an ink or dye image to an ink or dye-receiving or recording element as described above to form the ink or dye transfer image. The thermal ink or dye transfer method of printing, an ink or dye donor element may be employed which comprises a poly(ethylene terephthalate) support coated with sequential repeating areas of cyan, magenta, and yellow ink or dye, and the ink or dye transfer steps may be sequentially performed for each color to obtain a three-color ink or dye transfer image. When the process is only performed for a single color, then a monochrome ink or dye transfer image may be obtained.

[0064] Thermal printing heads which can be used to transfer ink or dye from ink or dye-donor elements to receiving or recording elements of the invention are available commercially. There can be employed, for example, a Fujitsu Thermal Head (FTP-040 MCS001), a TDK Thermal Head F415 HH7-1089, or a Rohm Thermal Head KE 2008-F3. Alternatively, other known sources of energy for thermal ink or dye transfer may be used, such as lasers as described in, for example, GB No. 2,083,726A.

[0065] A thermal ink or dye transfer assemblage may comprise (a) an ink or dye-donor element, and (b) an ink or dye-receiving or recording element as described above, the ink or dye-receiving or recording element being in a superposed relationship with the ink or dye-donor element so that the ink or dye layer of the donor element may be in contact with the ink or dye image-receiving or recording layer of the receiving or recording element.

[0066] When a three-color image is to be obtained, the above assemblage may be formed on three occasions during the time when heat is applied by the thermal printing head. After the first dye is transferred, the elements may be peeled apart. A second dye-donor element (or another area of the donor element with a different dye area) may be then brought in register with the dye-receiving or recording element and the process repeated. The third color may be obtained in the same manner.

[0067] The electrographic and electrophotographic processes and their individual steps have been well described in the prior art. The processes incorporate the basic steps of creating an electrostatic image, developing that image with charged, colored particles (toner), optionally transferring the resulting developed image to a secondary substrate, and fixing the image to the substrate. There are numerous variations in these processes and basic steps, the use of liquid toners in place of dry toners is simply one of those variations.

[0068] The first basic step, creation of an electrostatic image, can be accomplished by a variety of methods. In one form, the electrophotographic process of copiers uses imagewise photodischarge, through analog or digital exposure, of a uniformly charged photoconductor. The photoconductor may be a single-use system, or it may be rechargeable and reimageable, like those based on selenium or organic photoreceptors.

[0069] In an alternate electrographic process, electrostatic images may be created ionographically. The latent image may be created on dielectric (charge-holding) medium, either paper or film. Voltage may be applied to selected metal styli or writing nibs from an array of styli spaced across the width of the medium, causing a dielectric breakdown of the air between the selected styli and the medium. Ions may be created, which form the latent image on the medium.

[0070] Electrostatic images, however generated, may be developed with oppositely charged toner particles. For development with liquid toners, the liquid developer may be brought into direct contact with the electrostatic image. Usually a flowing liquid is employed to ensure that sufficient toner particles are available for development. The field created by the electrostatic image causes the charged particles, suspended in a nonconductive liquid, to move by electrophoresis. The charge of the latent electrostatic image may be thus neutralized by the oppositely charged particles. The theory and physics of electrophoretic development with liquid toners are well described in many books and publications.

[0071] If a reimageable photoreceptor or an electrographic master is used, the toned image may be transferred to paper (or other substrate). The paper may be charged electrostatically, with the polarity chosen to cause the toner particles to transfer to the paper. Finally, the toned image may be fixed to the paper. For self-fixing toners, residual liquid may be removed from the paper by air-drying or heating. Upon evaporation of the solvent, these toners form a film bonded to the paper. For heat-fusible toners, thermoplastic polymers may be used as part of the particle. Heating both removes residual liquid and fixes the toner to paper.

[0072] When used as inkjet imaging media, the recording elements or media typically comprise a substrate or a support material having on at least one surface thereof an ink-receiving or recording / recording or image-forming layer. If desired, in order to improve the adhesion of the inkjet receiving or recording layer to the support, the surface of the support may be corona-discharge-treated prior to applying the solvent-absorbing layer to the support or, alternatively, an undercoating, such as a layer formed from a halogenated phenol or a partially hydrolyzed vinyl chloride-vinyl acetate copolymer, can be applied to the surface of the support. The inkjet receiving or recording layer is preferably coated onto the support layer from water or water-alcohol solutions at a dry thickness ranging from 3 to 75 micrometers,

preferably 8 to 50 micrometers.

[0073] Any known inkjet receiver layer can be used in combination with other particulate materials. For example, the ink receiving or recording layer may consist primarily of inorganic oxide particles such as silicas, modified silicas, clays, aluminas, fusible beads such as beads comprised of thermoplastic or thermosetting polymers, non-fusible organic beads, or hydrophilic polymers such as naturally-occurring hydrophilic colloids and gums such as gelatin, albumin, guar, xanthan, acacia, chitosan, starches and their derivatives, derivatives of natural polymers such as functionalized proteins, functionalized gums and starches, and cellulose ethers and their derivatives, and synthetic polymers such as polyvinylloxazoline, polyvinylmethyloxazoline, polyoxides, polyethers, poly(ethylene imine), poly(acrylic acid), poly(methacrylic acid), n-vinyl amides including polyacrylamide and polyvinylpyrrolidone, and poly(vinyl alcohol), its derivatives and copolymers, and combinations of these materials. Hydrophilic polymers, inorganic oxide particles, and organic beads may be present in one or more layers on the substrate and in various combinations within a layer.

[0074] A porous structure may be introduced into ink receiving or recording layers comprised of hydrophilic polymers by the addition of ceramic or hard polymeric particulates, by foaming or blowing during coating, or by inducing phase separation in the layer through introduction of non-solvent. In general, it is preferred for the base layer to be hydrophilic, but not porous. This may be especially true for photographic quality prints, in which porosity may cause a loss in gloss. In particular, the ink receiving or recording layer may consist of any hydrophilic polymer or combination of polymers with or without additives as is well known in the art.

[0075] If desired, the ink receiving or recording layer can be overcoated with an ink-permeable, anti-tack protective layer such as, for example, a layer comprising a cellulose derivative or a cationically-modified cellulose derivative or mixtures thereof. An especially preferred overcoat is poly(1,4-anhydro-glucose-g-oxyethylene-g-(2'-hydroxypropyl)-N, N-dimethyl-N-dodecylammonium chloride). The overcoat layer is non porous, but is ink permeable and serves to improve the optical density of the images printed on the element with water-based inks. The overcoat layer can also protect the ink receiving or recording layer from abrasion, smudging, and water damage. In general, this overcoat layer may be present at a dry thickness of 0.1 to 5 μm , preferably 0.25 to 3 μm .

[0076] In practice, various additives may be employed in the ink receiving or recording layer and overcoat. These additives include surface active agents such as surfactant(s) to improve coatability and to adjust the surface tension of the dried coating, acid or base to control the pH, antistatic agents, suspending agents, antioxidants, hardening agents to cross-link the coating, antioxidants, UV stabilizers, light stabilizers, and thickeners. In addition, a mordant may be added in small quantities (2%-10% by weight of the base layer) to improve waterfastness. Useful mordants are disclosed in U.S. Patent No. 5,474,843.

[0077] The layers described above, including the ink receiving or recording layer and the overcoat layer, may be coated by conventional coating means onto a transparent or opaque support material commonly used in this art. Coating methods may include, but are not limited to, blade coating, wound wire rod coating, slot coating, slide hopper coating, gravure, curtain coating. Some of these methods allow for simultaneous coatings of both layers, which is preferred from a manufacturing economic perspective.

[0078] The IRL (ink or dye receiving layer) may be coated over a tie layer (TL). There are many known formulations, which may be useful as ink or dye receiving or recording layers. The primary requirement is that the IRL may be compatible with the inks which it will be imaged so as to yield the desirable color gamut and density. As the ink drops pass through the IRL, the ink or dyes may be retained or mordanted in the IRL, while the ink solvents pass freely through the IRL and may be rapidly absorbed by the TL. Additionally, the IRL formulation may be preferably coated from water, exhibits adequate adhesion to the TL, and allows for easy control of the surface gloss.

[0079] For example, Misuda et al in US Patents 4,879,166, 5,264,275, 5,104,730, 4,879,166, and Japanese Patents 1,095,091, 2,276,671, 2,276,670, 4,267,180, 5,024,335, and 5,016,517 disclose aqueous based IRL formulations comprising mixtures of pseudo-bohemite and certain water soluble resins. Light in US Patents 4,903,040, 4,930,041, 5,084,338, 5,126,194, 5,126,195, and 5,147,717 discloses aqueous-based IRL formulations comprising mixtures of vinyl pyrrolidone polymers and certain water-dispersible and/or water-soluble polyesters, along with other polymers and addenda. Butters et al in US Patents 4,857,386 and 5,102,717 disclose ink-absorbent resin layers comprising mixtures of vinyl pyrrolidone polymers and acrylic or methacrylic polymers. Sato et al in US Patent 5,194,317 and Higuma et al in US Patent 5,059,983 disclose aqueous-coatable IRL formulations based on poly(vinyl alcohol). Iqbal in US Patent 5,208,092 discloses water-based IRL formulations comprising vinyl copolymers, which may be subsequently cross-linked. In addition to these examples, there may be other known or contemplated IRL formulations, which are consistent with the aforementioned primary and secondary requirements of the IRL, all of which fall under the spirit and scope of the current invention.

[0080] The IRL may also contain varying levels and sizes of matting agents for the purpose of controlling gloss, friction, and/or fingerprint resistance, surfactants to enhance surface uniformity and to adjust the surface tension of the dried coating, mordanting agents, antioxidants, UV absorbing compounds, light stabilizers.

[0081] It may also be desirable to overcoat the IRL for the purpose of enhancing the durability of the imaged element. Such overcoats may be applied to the IRL either before or after the element is imaged. For example, the IRL can be

overcoated with an ink-permeable layer through which inks freely pass. Layers of this type are described in US Patents 4,686,118, 5,027,131, and 5,102,717. Alternatively, an overcoat may be added after the element is imaged. Any of the known laminating films and equipment may be used for this purpose. The inks used in the aforementioned imaging process are well known, and the ink formulations are often closely tied to the specific processes, i.e., continuous, piezoelectric, or thermal. Therefore, depending on the specific ink process, the inks may contain widely differing amounts and combinations of solvents, colorants, preservatives, surfactants, humectants. Inks preferred for use in combination with the image recording elements of the present invention are water-based. However, it may be intended that alternative embodiments of the image-recording elements as described above, which may be formulated for use with inks which may be specific to a given ink-recording process or to a given commercial vendor, fall within the scope of the present invention.

[0082] In another embodiment, in order to produce photographic elements, the composite support sheet may be coated with a photographic element or elements. The photographic elements can be single color elements or multicolor elements. Multicolor elements contain image ink or dye-forming units sensitive to each of the three primary regions of the spectrum. Each unit can comprise a single emulsion layer or multiple emulsion layers sensitive to a given region of the spectrum. The layers of the element, including the layers of the image-forming units, can be arranged in various orders as known in the art. In an alternative format, the emulsions sensitive to each of the three primary regions of the spectrum can be disposed as a single segmented layer.

[0083] The photographic emulsions useful for this invention may be generally prepared by precipitating silver halide crystals in a colloidal matrix by methods conventional in the art. The colloid may be typically a hydrophilic film forming agent such as gelatin, alginic acid, or derivatives thereof.

[0084] The crystals formed in the precipitation step are washed and then chemically and spectrally sensitized by adding spectral sensitizing dyes and chemical sensitizers, and by providing a heating step during which the emulsion temperature may be raised, typically from 40.degree. C. to 70.degree. C., and maintained for a period of time. The precipitation and spectral and chemical sensitization methods utilized in preparing the emulsions employed in the invention can be those methods known in the art.

[0085] Chemical sensitization of the emulsion typically employs sensitizers such as: sulfur-containing compounds, e.g., allyl isothiocyanate, sodium thiosulfate and allyl thiourea, reducing agents, e.g., polyamines and stannous salts, noble metal compounds, e.g., gold, platinum, and polymeric agents, e.g., polyalkylene oxides. As described, heat treatment may be employed to complete chemical sensitization. Spectral sensitization may be effected with a combination of dyes, which are designed for the wavelength range of interest within the visible or infrared spectrum. It is known to add such dyes both before and after heat treatment.

[0086] After spectral sensitization, the emulsion may be coated on a support. Various coating techniques include dip coating, air knife coating, curtain coating and extrusion coating.

[0087] The silver halide emulsions utilized in this invention may be comprised of any halide distribution. Thus, they may be comprised of silver chloride, silver bromide, silver bromochloride, silver chlorobromide, silver iodochloride, silver iodobromide, silver bromiodochloride, silver chloriodobromide, silver iodobromochloride, and silver iodochlorobromide emulsions. By predominantly silver chloride, it is meant that the grains of the emulsion are greater than 50 mole percent silver chloride. Preferably, they are greater than 90 mole percent silver chloride, and optimally greater than 95 mole percent silver chloride.

[0088] The silver halide emulsions can contain grains of any size and morphology. Thus, the grains may take the form of cubes, octahedrons, cubooctahedrons, or any of the other naturally occurring morphologies of cubic lattice type silver halide grains. Further, the grains may be irregular such as spherical grains or tabular or core / shell grains. Grains having a tabular or cubic morphology are preferred.

[0089] The photographic elements used with the invention may utilize emulsions as described in The Theory of the Photographic Process, Fourth Edition, T. H. James, Macmillan Publishing Company, Inc., 1977, pages 151-152. Reduction sensitization has been known to improve the photographic sensitivity of silver halide emulsions. While reduction sensitized silver halide emulsions generally exhibit good photographic speed, they often suffer from undesirable fog and poor storage stability.

[0090] Reduction sensitization can be performed intentionally by adding reduction sensitizers, chemicals that reduce silver ions to form metallic silver atoms, or by providing a reducing environment such as high pH (excess hydroxide ion) and/or low pAg (excess silver ion). During precipitation of a silver halide emulsion, unintentional reduction sensitization can occur when, for example, silver nitrate or alkali solutions may be added rapidly or with poor mixing to form emulsion grains. Also, precipitation of silver halide emulsions in the presence of ripeners (grain growth modifiers) such as thioethers, selenoethers, thioureas, or ammonia tends to facilitate reduction sensitization.

[0091] Examples of reduction sensitizers and environments which may be used during precipitation or spectral/chemical sensitization to reduction sensitize an emulsion include ascorbic acid derivatives, tin compounds, polyamine compounds, and thiourea dioxide-based compounds described in U.S. Pat. Nos. 2,487,850, 2,512,925, and British Patent 789,823. Specific examples of reduction sensitizers or conditions, such as dimethylamineborane, stannous chloride,

hydrazine, high pH (pH 8-11) and low pAg (pAg 1-7) ripening are discussed by S. Collier in Photographic Science and Engineering, 23, 113 (1979). Examples of processes for preparing intentionally reduction sensitized silver halide emulsions are described in EP 0 348 934 Al (Yamashita), EP 0 369 491 (Yamashita), EP 0 371 388 (Ohashi), EP 0 396 424 Al (Takada), EP 0 404 142 Al (Yamada), and EP 0 435 355 Al (Makino).

[0092] The photographic elements of this invention may use emulsions doped with Group VII metals such as iridium, rhodium, osmium, and iron as described in Research Disclosure, September 1994, Item 36544, Section I, published by Kenneth Mason Publications, Ltd., Dudley Annex, 12a North Street, Emsworth, Hampshire PO10 7DQ, ENGLAND. Additionally, a general summary of the use of iridium in the sensitization of silver halide emulsions is contained in Carroll, "Iridium Sensitization: A Literature Review," Photographic Science and Engineering, Vol. 24, No. 6, 1980. A method of manufacturing a silver halide emulsion by chemically sensitizing the emulsion in the presence of an iridium salt and a photographic spectral sensitizing dye is described in U.S. Pat. No. 4,693,965. In some cases, when such dopants may be incorporated, emulsions show an increased fresh fog and a lower contrast sensitometric curve when processed in the color reversal E-6 process as described in The British Journal of Photography Annual, 1982, pages 201-203.

[0093] A typical multicolor photographic element used with the invention comprises the invention laminated support bearing a cyan ink or dye image-forming unit comprising at least one red-sensitive silver halide emulsion layer having associated therewith at least one cyan dye-forming coupler, a magenta image-forming unit comprising at least one green-sensitive silver halide emulsion layer having associated therewith at least one magenta dye-forming coupler, and a yellow dye image-forming unit comprising at least one blue-sensitive silver halide emulsion layer having associated therewith at least one yellow dye-forming coupler. The element may contain additional layers, such as filter layers, interlayers, overcoat layers, subbing layers. The support used with the invention may also be utilized for black and white photographic print elements.

[0094] The photographic elements may also contain a transparent magnetic recording layer such as a layer containing magnetic particles on the underside of a transparent support, as in U.S. Pat. Nos. 4,279,945 and 4,302,523. The invention may be utilized with the materials disclosed in Research Disclosure, September 1997, Item 40145. The invention may be particularly suitable for use with the material color paper examples of sections XVI and XVII. The couplers of section II are also particularly suitable. The Magenta I couplers of section II, particularly M-7, M-10, M-18, and M-18, set forth below are particularly desirable. In the following Table, reference will be made to (1) Research Disclosure, December 1978, Item 17643, (2) Research Disclosure, December 1989, Item 308119, and (3) Research Disclosure, September 1994, Item 36544, all published by Kenneth Mason Publications, Ltd., Dudley Annex, 12a North Street, Emsworth, Hampshire PO10 7DQ, ENGLAND. The Table and the references cited in the Table are to be read as describing particular components suitable for use in the elements of the invention. The Table and its cited references also describe suitable ways of preparing, exposing, processing and manipulating the elements and the images contained therein.

Reference Section	Subject Matter
1 I, II	Grain composition,
2 I, II, IX, X, XI,	morphology and preparation.
XII, XIV, XV	Emulsion preparation including
1, II, III, IX	hardeners, coating aids,
3 A & B	addenda, etc.
1 III, IV	Chemical sensitization and
2 III, IV	pectral sensitization
3 IV, V	Desensitization.
1 V	UV dyes, optical brighteners,
2 V	luminescent dyes
3 VI	
1 VI	Antifoggants and stabilizers
2 VI	
3 VII	
1 VIII	Absorbing and scattering
2 VIII, XIII, XVI	materials; Antistatic layers;
3 VIII, IX	C & D matting agents
1 VII	Image-couplers and image-
2 VII	modifying couplers; Dye

(continued)

Reference Section	Subject Matter
3 X	stabilizers and hue modifiers
1 XVII	Supports
2 XVII	
3 XV	
3 XI	Specific layer arrangements
3 XII, XIII	Negative working emulsions; Direct positive emulsions
2 XVIII	Exposure
3 XVI	
1 XIX, XX	Chemical processing;
2 XIX, XX, XXII	Developing agents
3 XVIII, XIX, XX	
3 XIV	Scanning and digital processing procedures

[0095] The photographic elements can be exposed with various forms of energy which encompass the ultraviolet, visible, and infrared regions of the electromagnetic spectrum as well as with electron beam, beta radiation, gamma radiation, x-ray, alpha particle, neutron radiation, and other forms of corpuscular and wave-like radiant energy in either noncoherent (random phase) forms or coherent (in phase) forms, as produced by lasers. When the photographic elements are intended to be exposed by x-rays, they can include features found in conventional radiographic elements.

[0096] The photographic elements may be preferably exposed to actinic radiation, typically in the visible region of the spectrum, to form a latent image, and then processed to form a visible image, preferably by other than heat treatment. Processing may be preferably carried out in the known RA-4.TM. (Eastman Kodak Company) Process or other processing systems suitable for developing high chloride emulsions. This invention may be also directed towards a photographic recording element comprising a support and at least one light sensitive silver halide emulsion layer comprising silver halide grains as described above.

[0097] Although the recording elements disclosed herein have been referred to primarily as being useful for inkjet printers, they also can be used as recording media for pen plotter assemblies. Pen plotters operate by writing directly on the surface of a recording medium using a pen consisting of a bundle of capillary tubes in contact with an ink reservoir.

[0098] The following examples are intended to further illustrate, but not to limit, the invention.

EXAMPLES

Preparation of porous polyester particles PE-1 through PE-10

[0099] Precursor polyesters PP-1 through PP-5 were all synthesized by a 2-stage melt polycondensation process. The chemical compositions are listed in Table 1 and the reaction times and final molecular weights are listed in Table 2. The SIP, CHDM, dibutylstannoic acid, zinc acetate, and sodium acetate were combined in a 500 ml 3-neck flask equipped with a stainless steel stirring rod, nitrogen inlet, and an arm leading to a dry ice/acetone condenser with an outlet connected to a controlled vacuum system. A graduated cylinder was connected beneath the condenser with a ground glass joint to collect and measure distillate. The reaction was heated in thermostatted bath containing a metal heating alloy. A steady stream of nitrogen was passed over the reaction mixture for 10 minutes, and was then reduced to a slightly positive flow. The temperature was held at 210-230°C for 100-250 minutes with stirring at ~100 RPM until a clear prepolymer resulted and the calculated amount of methanol distillate had been collected in a graduated cylinder. The reaction was removed from the bath and allowed to cool. The FA and IPA were then added and the condenser was filled with ice. The reaction was restarted at 220°C and within 20 minutes water condensate began to collect. The reaction was held at 220°C for 130-460 minutes until the viscosity of the melt had increased to the point where the precursor polyester could no longer be effectively stirred and the reaction was terminated.

Table 1

	FA (g)	SIP (g)	IPA (g)	Mol ratio FA: SIP:IPA	CHDM (g)	Zn(OAc) ₂ (g)	BuSn(OH) ₃ (g)	NaOAc (g)
PP-1	19.59	50.01	-	1:1:0	45.57	0.02	0.02	0.69
PP-2	21.43	27.35	15.34	2:1:1	51.13	0.02	0.02	0.38
PP-3	22.49	14.35	24.14	4:1:3	53.76	0.02	0.02	0.20
PP-4	20.49	39.23	7.33	4:3:1	48.80	0.02	0.20	0.54
PP-5	48.99	125.02	-	1:1:0	116.42	0.06	0.04	1.73
FA is fumaric acid SIP is dimethyl 5-sulfoisophthalate, sodium salt IPA is isophthalic acid CHDM is 1,4-cyclohexanedimethanol, mixture of <i>cis</i> and <i>trans</i> isomers.								

Table 2

	Stage 1 reaction time	Stage 2 reaction time	Mn	Mw
PP-1	120 min	130 min	2,710	10,500
PP-2	100 min	420 min	3,570	15,800
PP-3	100 min	460 min	3,380	28,000
PP-4	120 min	180 min	3,130	7,340
PP-5	250 min	200 min	3,240	7,330

[0100] Precursor polyester PP-6 was synthesized using the same apparatus as used in the preparation of PP-1 through PP-5, and a similar 2-stage procedure. In the first stage, SIP (82.41 g, 0.28 mol), hydroquinone-bis-hydroxyethyl ether (HQBHE) (110.29 g, 0.56 mol), sodium acetate (2.28 g), and titanium isopropoxide (4 drops) were combined in a 500 ml 3-neck round bottom flask and heated for 220-240°C for 70 minutes at which point a clear, slightly orange prepolymer had formed and the calculated amount of methanol condensate had been collected. The reaction was then cooled and diethyl fumarate (47.90 g, 0.28 mol) was added. The reaction was restarted and was heated at 200°C for 100 minutes followed by 120 minutes at 220°C. A vacuum was then initiated which ramped from ambient pressure to 20 torr over 2 minutes. The viscosity of the melt began to rapidly increase and the reaction was terminated. (Mn = 2940, Mw = 5440).

[0101] Polyester beads PE-1 through PE-10 were all synthesized by the following procedure. An aqueous phase was prepared by dispersing appropriate precursor polyester in the amount of water noted in Table 3. The water generally had to be heated to 40-60°C and it required from 20 minutes to 16 hours for the precursor polyester to completely disperse, depending on the amount of SIP monomer in the precursor polyester. The aqueous phase was filtered through cheesecloth and combined in a beaker with an organic phase consisting of the toluene, DVB, hexadecane, and AIBN. The 2 phases were emulsified by any of the three methods listed in Table 3 and transferred to an appropriately sized 3-neck round bottom flask (1 L for PE-1 through 8 or 2 L for PE-9 through 10) fitted with a mechanical stirrer and a reflux condenser with nitrogen inlet. The opaque white microsuspensions were bubble degassed with nitrogen for 10 minutes, then heated overnight at 70°C for 16 hours. The resulting particle dispersions were cooled to room temperature and the toluene was removed as a water azeotrope via rotary evaporation. The dispersions were washed with 4-6 volumes of water and concentrated to 9-25% solids using a Millipore Amicon ultrafiltration system with a 100K cutoff spiral-wound dialysis cartridge. The exact concentrations of each PE dispersion are listed in Table 4.

Table 3

Disper sion	Precursor polyester	Precursor polyester (g)	Water (ml)	DVB ¹ (g)	Toluene (g)	AIBN ⁵ (g)	hexadecane (ml)	Emulsification method
PE-1	PP-1	20.00	240	20.00	40.00	0.40	4.14	M ²
PE-2	PP-2	20.00	240	20.00	40.00	0.40	4.14	M2
PE-3	PP-3	20.00	240	20.00	40.00	0.40	4.14	M2
PE-4	PP-4	20.00	240	20.00	40.00	0.40	4.14	S ³
PE-5	PP-2	20.00	240	20.00	40.00	0.40	4.14	S3
PE-6	PP-3	20.00	240	20.00	40.00	0.40	4.14	S3
PE-7	PP-1	20.00	240	20.00	40.00	0.40	4.14	H ⁴
PE-8	PP-3	20.00	240	20.00	40.00	0.40	4.14	H4
PE-9	PP-5	62.50	750	68.38	125.00	1.25	12.94	H4
PE-10	PP-6	40.00	514	40.0	80.0	0.80	2.82	H4

¹ DVB is divinylbenzene (80% w/w with remainder being ethylstyrene, mixture of *m* and *p* isomers.)

² Reaction mixture was passed twice through an M-110T Microfluidizer (sold by Microfluidics).

³ Reaction mixture was homogenized for 10 minutes using a Silverson L4R mixer at the highest speed setting and then sonicated using a Vibra Cell probe sonicator (Sonics & Materials Inc.).

⁴ Reaction mixture was homogenized using a Silverson L4R mixer at the highest speed setting for 10 minutes.

⁵ AIBN is 2,2'-azobis(isobutyronitrile)

Determination of particle size and particle size distribution

[0102] The particle size and particle size distribution of each PE-1 through PE-10 was measured using a laser scattering particle size distribution analyzer, Horiba LA-920, manufactured by Horiba LTD. The results are in Table 4.

Table 4

PE dispersion	Wt % solids	Mode 1			Mode 2		
		Mean diameter (micron)	Proportion (%)	CV (%)	Mean diameter (micron)	Proportion (%)	CV %
PE-1	13.54	0.356	100	35.2	--	--	--
PE-2	12.97	0.181	5.9	15.9	0.351	94.1	36.8
PE-3	14.01	0.390	88.2	41.9	2.70	11.8	38.1
PE-4	14.37	0.470	58.6	55.3	2.24	41.4	25.2
PE-5	12.38	0.174	45.1	21.9	0.391	54.9	31.9
PE-6	15.50	0.430	82.1	42.2	5.86	17.9	54.0
PE-7	14.00	0.897	10.6	27.1	2.62	89.4	36.0
PE-8	16.24	0.682	7.6	36.4	3.10	92.4	37.3
PE-9	22.86	1.082	9.1	44.2	2.69	90.9	27.1
PE-10	17.60	0.434	27.6	45.3	4.46	72.4	57.5

3-step Preparation of Porous Beads Containing Polyester-carbonate (PEC-1)

[0103] Polyester-diol prepolymer: Fumaric acid (100.00 g, 6.61×10^{-1} mol), cyclohexanedimethanol (174.04 g, 1.20 mol), and Fascat 4100 (~ 10 mg, catalytic) were combined in a 500 ml single neck round bottom flask outfitted with a

mechanical stirrer and a side arm leading to a condenser assembly with a graduated cylinder to measure the volume of condensate. The reaction was placed in a bath filled with a low melting heating alloy and the temperature was ramped from 150°C to 220°C over 50 minutes then held at 220°C for 200 minutes at which point ~32 ml of condensate had been collected. The molten polyester was cooled to room temperature, frozen in liquid nitrogen, broken up with a hammer and dried in a vacuum oven overnight at 60°C. The alcohol end groups were determined by dissolving the polyester in deuteriochloroform, endcapping the polyester with trifluoroacetylimidazole, integrating the ¹⁹F NMR end group peaks, and normalizing to an internal standard. The determined value of 2.28 meq ROH/g polymer corresponds to a number average molecular weight of 877.19 g/mol.

[0104] Unsaturated polyester-carbonate: The unsaturated polyester diol oligomer from the previous step (43.85 g, 0.05 mol) and trimethylamine (22.26 g, 0.22 mol) were dissolved in dichloromethane (150 g). While stirring at room temperature, a solution of bisphenol A bis(chloroformate) (47.66 g, 0.05 g) in dichloromethane (50 ml) was slowly added over ~20 min. The reaction was allowed to proceed at room temperature for two hours, by which time it had become notably viscous and was further diluted 100 g dichloromethane. The product solution was then extracted once with 250 ml of 2% hydrochloric acid, and twice with water. The product solution was precipitated into methanol and dried overnight in a vacuum oven at 60°C. Analysis by Differential scanning calorimetry showed a T_g of 89.4°C. Size exclusion chromatography in hexafluoroisopropanol showed M_n = 1,390 and M_w = 80,700. The broad curve suggested that the polymer may be highly branched.

[0105] Particles containing polyester-co-carbonate: An organic phase was prepared consisting of the unsaturated polyester-co-carbonate from the previous step (10.00 g) toluene (20.00 g), chloromethylstyrene (5.00 g), 80% divinylbenzene (5.00g, mixture of *m* and *p* isomers with remainder being ethylstyrene), hexadecane (0.80 g), and Vazo 52® initiator (0.20 g). The components were stirred until a clear, heterogeneous solution resulted. An aqueous phase was prepared consisting of dodecanethiol-endcapped polyacrylamide decamer surfactant (1.60 g, prepared as described in US 6,127,453 column 9 lines 40-55) dissolved in deionized water (120.00). The two phases were combined and sonicated using a Vibra Cell® probe sonicator (Sonics & Materials Inc.) at the highest power setting for 10 minutes, then transferred to a 500 ml 3-neck round bottomed flask outfitted with a condenser, nitrogen inlet, and mechanical stirrer. The reaction mixture was bubble degassed with nitrogen for 10 minutes and heated for 16 hours with vigorous stirring at 60°C in a thermostatted water bath. The toluene was then removed via rotary evaporation and dimethylethanolamine (1.17 g) was added. The quaternization reaction was allowed to proceed at 60°C for 16 hours and the excess amine and surfactant was removed via dialysis using 12-14K cutoff cellulose dialysis tubing for 16 hours to yield a product dispersion of 13.3% solids. Analysis of the product dispersion using a Horiba® LA-920 particle sizing instrument showed a bimodal distribution with a mean particle diameter of 1.462 microns, a major mode at 0.36 microns, and a smaller mode, presumably due to secondary aggregation, at ~6 microns.

2-step Preparation of Bead Containing Polyester-urethane (PEU-1).

[0106] Polyester-diol prepolymer: This reaction was set up and run in an identical manner as that described in step 1 of the preparation of PEC-1 using Maleic anhydride (107.61 g, 1.10 mol), neopentyl glycol (137.15 g, 1.32 mol), and Fascat 4100 (~ 5 mg, catalytic). The alcohol end groups were determined by direct quantification using ¹H NMR with an internal standard. The determined value of 2.0 meq ROH/g polymer corresponds to a number average molecular weight of 1036.3 g/mol.

[0107] Particles containing polyester-urethane: An organic phase was prepared consisting of the polyester-diol oligomer from the previous step (10.00 g) toluene (55.20 g), Desmodur N3300® triisocyanate resin (3.80 g), stannous octanoate (0.05 g), dibutyltin dilaurate (0.05 g), 80% divinylbenzene (41.40g, mixture of *m* and *p* isomers with remainder being ethylstyrene), hexadecane (2.21 g), and AIBN initiator (1.03 g). The components were stirred until a clear, heterogeneous solution resulted. An aqueous phase was prepared consisting of sodium dodecylsulfate (4.40 g) dissolved in deionized water (120.00). The two phases were combined and emulsified using a Silverson L4R mixer at the highest speed for 10 minutes followed by passage twice through an M-110T Microfluidizer® (sold by Microfluidics). The resulting emulsion was poured into a 3-neck 2000 ml round bottom flask outfitted with a mechanical stirrer, condenser, and nitrogen inlet and was bubble degassed with nitrogen for 10 minutes. The emulsion was heated for 16 hours with vigorous stirring in a thermostatted water bath at 70°C and the toluene was removed via rotary evaporation. The resulting particle dispersion had 10.23% solids. Analysis of the product dispersion using a Horiba® LA-920 particle sizing instrument showed a single mode with a mean particle diameter of 0.153 microns.

Preparation of Element 1

[0108] A coating composition was prepared from 62.78 wt. % of dispersion PE-1, 1.5 wt. % poly(vinyl alcohol), PVA, (Gohsenol® GH-17 from Nippon Gohsei Co.), and 35.72 wt. % water. [The relative proportions of porous polyester particle to PVA are therefore 85/15 by weight]. The coating composition was coated onto a base support comprised of

a polyethylene resin coated photographic paper stock, which had been previously subjected to corona discharge treatment, using a calibrated coating knife, and dried to remove substantially all solvent components to form the ink receiving layer. The thickness of the dry ink receiving layer was measured to be about $17 \pm 2 \mu\text{m}$. The measured surface gloss at 60 degrees was 36 and the ink dry time was measured to be less than 10 seconds.

Preparation of Element 2

[0109] A coating composition was prepared from 61.68 wt. % of dispersion PE-2, 2.0 wt. % poly(vinyl alcohol), PVA, (Gohsenol® GH-17 from Nippon Gohsei Co.), and 36.32 wt. % water. [The relative proportions of porous polyester particle to PVA are therefore 80/20 by weight]. The coating composition was coated and dried the same as Element 1. The thickness of the dry ink receiving layer was measured to be about $19 \pm 2 \mu\text{m}$.

Preparation of Element 3

[0110] A coating composition was prepared from 57.10 wt. % of dispersion PE-3, 2.0 wt. % poly(vinyl alcohol), PVA, (Gohsenol® GH-17 from Nippon Gohsei Co.), and 40.90 wt. % water. [The relative proportions of porous polyester particle to PVA are therefore 80/20 by weight]. The coating composition was coated and dried the same as Element 1. The thickness of the dry ink receiving layer was measured to be about $46 \pm 2 \mu\text{m}$.

Preparation of Element 4

[0111] A coating composition was prepared from 59.15 wt. % of dispersion PE-4, 1.5 wt. % poly(vinyl alcohol), PVA, (Gohsenol® GH-17 from Nippon Gohsei Co.), and 39.35 wt. % water. [The relative proportions of porous polyester particle to PVA are therefore 85/15 by weight]. The coating composition was coated and dried the same as Element 1. The thickness of the dry ink receiving layer was measured to be about $18 \pm 2 \mu\text{m}$.

Preparation of Element 5

[0112] A coating composition was prepared from 64.62 wt. % of dispersion PE-5, 2.0 wt. % poly(vinyl alcohol), PVA, (Gohsenol® GH-17 from Nippon Gohsei Co.), and 33.38 wt. % water. [The relative proportions of porous polyester particle to PVA are therefore 80/20 by weight]. The coating composition was coated and dried the same as Element 1. The thickness of the dry ink receiving layer was measured to be about $25 \pm 2 \mu\text{m}$.

Preparation of Element 6

[0113] A coating composition was prepared from 51.61 wt. % of dispersion PE-6, 2.0 wt. % poly(vinyl alcohol), PVA, (Gohsenol® GH-17 from Nippon Gohsei Co.), and 46.39 wt. % water. [The relative proportions of porous polyester particle to PVA are therefore 80/20 by weight]. The coating composition was coated and dried the same as Element 1. The thickness of the dry ink receiving layer was measured to be about $24 \pm 2 \mu\text{m}$.

Preparation of Control Element C-1

[0114] A coating composition was prepared from 60.71 wt. % of dispersion PE-7, 1.5 wt. % poly(vinyl alcohol), PVA, (Gohsenol® GH-17 from Nippon Gohsei Co.), and 37.79 wt. % water. [The relative proportions of porous polyester particle to PVA are therefore 85/15 by weight]. The coating composition was coated and dried the same as Element 1. The thickness of the dry ink receiving layer was measured to be about $26 \pm 2 \mu\text{m}$.

Preparation of Control Element C-2

[0115] A coating composition was prepared from 52.34 wt. % of dispersion PE-8, 1.5 wt. % poly(vinyl alcohol), PVA, (Gohsenol® GH-17 from Nippon Gohsei Co.), and 46.16 wt. % water. [The relative proportions of porous polyester particle to PVA are therefore 85/15 by weight]. The coating composition was coated and dried the same as Element 1. The thickness of the dry ink receiving layer was measured to be about $32 \pm 2 \mu\text{m}$.

Preparation of Control Element C-3

[0116] Control Element C-3 was a commercially available inkjet non-porous receiver paper containing a high amount of gelatin, "Kodak Inkjet Photo Paper", catalogue No. 1181197 from Eastman Kodak Company.

Preparation of Control Element C-4

[0117] Control Element C-4 was a commercially available inkjet porous receiver paper containing a high amount of silica fine particles, "Epson Premium Glossy Photo Paper", catalogue No. SO41286 from Epson.

Printing and dye stability testing

[0118] The above elements and control elements of Example 1 were printed using a Lexmark Z51 inkjet printer and a cyan inkjet ink, prepared using a standard formulation with a copper phthalocyanine dye (Clariant Direct Turquoise Blue FRL-SF). The red channel density (cyan) patches at D-max (the highest density setting) were read using an X-Rite ® 820 densitometer. The cyan density is reported in Table 5. The printed elements were then subjected to 4 days exposure to a nitrogen flow containing 5 ppm ozone. The density of each patch was read after the exposure test using an X-Rite ® 820 densitometer. The % dye retention was calculated as the ratio of the density after the exposure test to the density before the exposure test. The results for cyan D-max are reported in Table 5.

Surface gloss measurement

[0119] The gloss of the top surface of the unprinted ink receiving layer was measured using a BYK Gardner gloss meter at an angle of illumination/reflection of 60°. The results are related to a highly polished black glass with a refractive index of 1.567 that has a specular gloss value of 100. The results are reported in Table 5.

Measurement of ink dry time:

[0120] A drop (from a 10 microliter capillary tube) of a magenta inkjet ink, prepared using a standard formulation with Dye 6 from U.S. Patent 6,001,161, was placed on each unprinted element and the time that it took for this spot to become dry to the touch was measured as the "ink drying time" as shown in Table 5.

Table 5

Element	Cyan D-max	% dye retention cyan D-max	60° gloss	Ink drying time (seconds)
1	1.6	91	36	< 10
2	1.7	94	72	60
3	1.6	96	20	< 10
4	1.6	85	28	40
5	1.7	96	70	150
6	1.8	99	40	< 10
Control C-1	1.1	96	3	< 10
Control C-2	1.4	93	5	50
Control C-3	1.8	99	88	> 180
Control C-4	2.2	27	38	45

[0121] The above results show that high surface gloss for the element containing porous polyester particles can only be achieved with the incorporation of porous polyester particles that have a mean diameter of less than 0.5 micrometers, compared to the control elements having only porous polyester particles that have a mean diameter of greater than 0.5 micrometers. The above results also show that, although high surface gloss can be attained with a layer that does not contain porous polyester particles (Control element C-3), rapid ink drying time is not achieved unless the layer contains porous polyester particles. The above results also show that, although high surface gloss and rapid ink drying time can be attained with a porous layer comprising fine-particle silica (Control Element C-4), dye stability towards environmental gases remains poor.

Example 2Preparation of Control Element C-5 (single layer of larger size particles)

[0122] A coating composition was prepared from 53.92 wt. % of dispersion PE-9, 2.18 wt. % poly(vinyl alcohol), PVA, (Gohsenol® GH-17 from Nippon Gohsei Co.), and 43.90 wt. % water. [The relative proportions of porous polyester particle to PVA are therefore 85/15 by weight]. The coating composition was metered to a slot-die coating apparatus and coated onto a base support comprised of a polyethylene resin coated photographic paper stock, which had been previously subjected to corona discharge treatment, moving at a speed of about 2.4 m/min. The coated support immediately entered the drying section of the coating machine to remove substantially all solvent components and form an ink receiving layer. The thickness of the dry ink receiving layer was measured to be about $15 \pm 2 \mu\text{m}$. The measured surface gloss at 60 degrees was 3 and the ink dry time was measured to be greater than 180 seconds.

Preparation of Element 7

[0123] A coating composition (Solution S-9) was prepared the same as in Control Element C-5. A second homogeneous coating composition (Solution S-1) was prepared the same as in Element 1. Both coating compositions were metered to a multiple-slot-die coating apparatus and coated simultaneously, with Solution S-9 being located below (closer to the support than) Solution S-1, onto a base support comprised of a polyethylene resin coated photographic paper stock, which had been previously subjected to corona discharge treatment, moving at a speed of about 2.4 m/min. The coated support immediately entered the drying section of the coating machine to remove substantially all solvent components and form a two-layer image receiving element. The combined thickness of the dry ink receiving layers was measured to be about $19 \pm 2 \mu\text{m}$. The measured surface gloss at 60 degrees was 23 and the ink dry time was measured to be 100 seconds.

Preparation of Element 8

[0124] Coating compositions S-9 and S-1 were prepared the same as Element 7 except that Solution S-1 was 59.08 wt. % of dispersion PE-1, 2.0 wt. % poly(vinyl alcohol), PVA, (Gohsenol® GH-17 from Nippon Gohsei Co.), and 38.92 wt. % water. [The relative proportions of porous polyester particle to PVA are therefore 80/20 by weight]. The coating compositions were coated and dried the same as Element 7. The combined thickness of the dry ink receiving layers was measured to be about $20 \pm 2 \mu\text{m}$. The measured surface gloss and the ink dry time are reported in Table 6.

Preparation of Element 9

[0125] A coating composition (Solution S-9) was prepared the same as in Control Element C-5. A second homogeneous coating composition (Solution S-2) was prepared the same as in Element 2 except that Solution S-2 was prepared from 65.54 wt. % of dispersion PE-2, 1.5 wt. % poly(vinyl alcohol), PVA, (Gohsenol® GH-17 from Nippon Gohsei Co.), and 32.96 wt. % water. [The relative proportions of porous polyester particle to PVA are therefore 85/15 by weight]. The coating compositions were coated and dried the same as Element 7, with Solution S-9 being located below (closer to the support than) Solution S-2. The thickness of the dry lower layer was measured to be about $12 \pm 2 \mu\text{m}$ and the thickness of the dry upper layer was measured to be about $9 \pm 2 \mu\text{m}$. The measured surface gloss and the ink dry time are reported in Table 6.

Preparation of Element 10

[0126] Coating compositions S-9 and S-2 were prepared, coated and dried the same as Element 9 except that the thickness of the layers were different. The combined thickness of the dry ink receiving layers was measured to be about $12 \pm 2 \mu\text{m}$. The measured surface gloss and the ink dry time are reported in Table 6.

Preparation of Element 11

[0127] A coating composition (Solution S-9) was prepared the same as in Control Element C-5. A second homogeneous coating composition (Solution S-10) was prepared from 48.30 wt. % of dispersion PE-10, 1.5 wt. % poly(vinyl alcohol), PVA, (Gohsenol® GH-17 from Nippon Gohsei Co.), and 50.20 wt. % water. [The relative proportions of porous polyester particle to PVA are therefore 85/15 by weight]. The coating compositions were coated and dried the same as Element 7, with Solution S-9 being located below (closer to the support than) Solution S-10. The thickness of the dry lower layer was measured to be about $20 \pm 2 \mu\text{m}$ and the thickness of the dry upper layer was measured to be

about $5 \pm 2 \mu\text{m}$. The measured surface gloss and the ink dry time are reported in Table 6.

[0128] The above elements of Example 2 were printed using a Lexmark Z51 inkjet printer and a cyan inkjet ink, prepared using a standard formulation with a copper phthalocyanine dye (Clariant Direct Turquoise Blue FRL-SF). The red channel density (cyan) patches at D-max (the highest density setting) are reported in Table 6.

Table 6

Element	Cyan D-max	60° gloss	Ink drying time (seconds)
7	1.7	23	100
8	1.6	30	120
9	1.7	65	120
10	1.7	48	180
11	1.6	10	20
Control C-5	1.1	3	> 180

[0129] The above results show that high surface gloss and improved print density for the element can be achieved with a multi-layer ink receiving layer structure, particularly when the upper layer contains porous polyester particles that have a mean diameter of less than 0.5 micrometers.

Example 3

Preparation of Element 12

[0130] A coating composition was prepared from 48.7 wt. % of aqueous dispersion PEU-1 (10.23 wt % solids in water), 5.0 wt. % poly(vinyl alcohol), PVA, (Gohsenol® GH-17 from Nippon Gohsei Co.), and 46.3 wt. % water. [The relative proportions of porous organic particle to PVA are therefore 50/50 by weight]. The coating composition was coated and dried the same as Element 1. Preparation of Element 13

[0131] A coating composition was prepared from 38.0 wt. % of aqueous dispersion PEC-1 (13.33 wt % solids in water), 5.0 wt. % poly(vinyl alcohol), PVA, (Gohsenol® GH-17 from Nippon Gohsei Co.), and 57.0 wt. % water. [The relative proportions of porous polyester particle to PVA are therefore 50/50 by weight]. The coating composition was coated and dried the same as Element 1. The thickness of the dry ink receiving layer was measured to be about $11 \pm 2 \mu\text{m}$.

[0132] The above elements of Example 3 were printed using a Lexmark Z51 inkjet printer and a cyan inkjet ink, prepared using a standard formulation with a copper phthalocyanine dye (Clariant Direct Turquoise Blue FRL-SF). The red channel density (cyan) patches at D-max (the highest density setting) were measured to be 1.5 for element 12 and 1.7 for element 13.

Claims

1. An imaging element comprising at least one layer comprising porous organic particles, wherein said porous organic particles comprise an unsaturated condensation polymer.
2. The imaging element of claim 1 wherein said porous organic particles comprise an unsaturated condensation polymer reacted with a vinyl monomer.
3. The imaging element of claims 1-2 wherein said unsaturated condensation polymer comprises ionic groups having quaternary ammonium moieties.
4. The imaging element of claim 1 wherein said porous organic particles comprise porous polyester particles having a mean diameter of less than 0.5 micrometers.
5. The imaging element of claim 4 wherein said particles have an ionic group equivalent weight of from 40 to 2000 grams per mole of ionic unit.

6. The imaging element of claims 4-5 wherein said porous polyester particles are a component of a system including at least one particle size system distribution mode of a mean particle diameter greater than 0.5 micrometers.
7. The imaging element of claim 4-6 wherein said porous particles are a component of a system including at least one particle size system distribution mode of a mean particle diameter greater than 0.5 micrometers.
8. The imaging element of claim 1-7 wherein said imaging element comprises an ink recording element, wherein said ink recording element comprises a support having thereon an ink receiving layer capable of accepting an ink image comprising at least one layer comprising porous polyester particles, wherein said porous polyester particles comprises a mean diameter of less than 0.5 micrometers.
9. The imaging element of claim 8 wherein said at least one layer further comprises a polymeric binder.
10. The imaging element of claim 1-9 wherein said particles comprise from 50 to 95% by weight of said at least one layer.